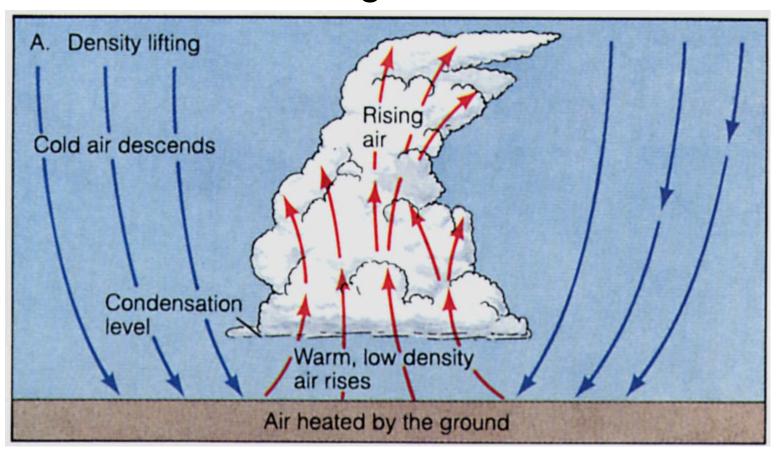
CE 374 K – Hydrology

Atmospheric Water

Daene C. McKinney

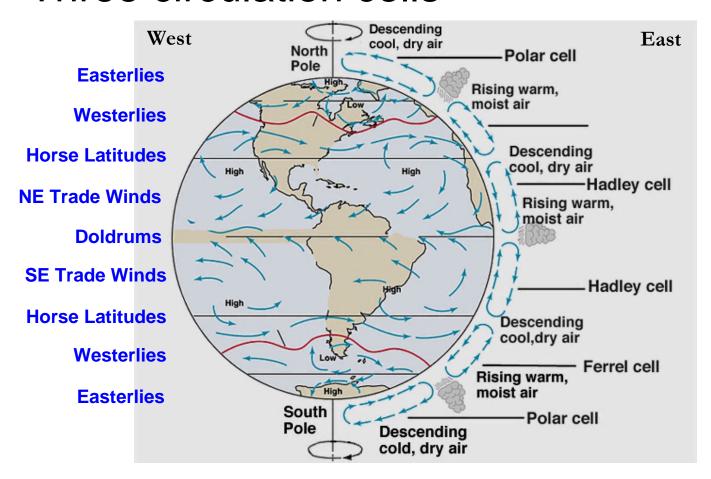
Heat Transport

Differential heating causes circulation

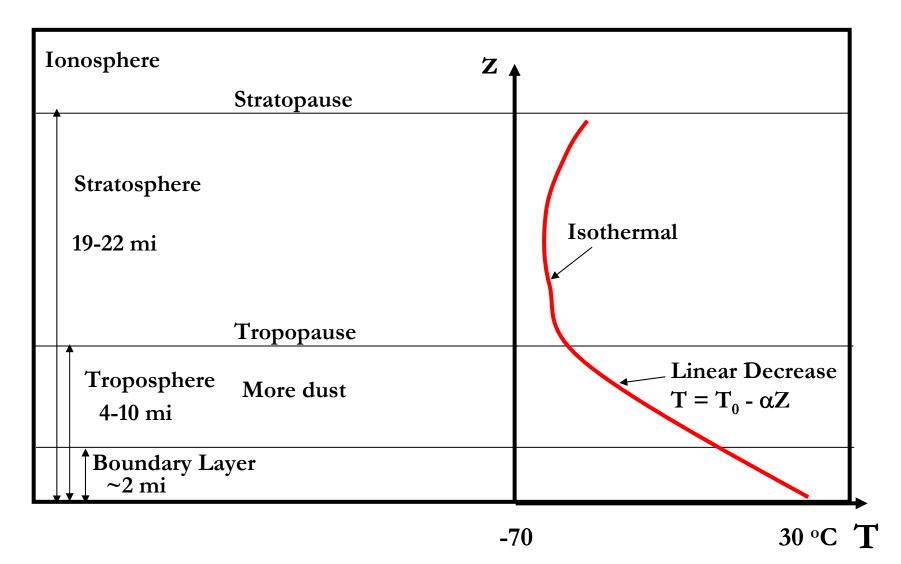


Rotating Earth

Three circulation cells



Structure of the Atmosphere



Temperature and Pressure in the Atmosphere

Troposphere

- Temperature decrease $T(Z) = T_0 \alpha Z$ α = lapse rate = 6.5 °C/km in standard atmosphere
- Pressure distribution

$$p(Z) = p_0 \left(\frac{T}{T_0}\right)^{g/R\alpha}$$

Stratosphere

Pressure distribution

$$p(Z) = p_0 \exp\left(-\frac{gZ}{RT}\right)$$

Composition of the Atmosphere

Components in Dry Air	Volume Ratio compared to Dry Air	Molecular Mass - <i>M</i> (kg/kmol)	Molecular Mass in Air
Oxygen	0.2095	32.00	6.704
Nitrogen	0.7809	28.02	21.88
Carbon Dioxide	0.0003	44.01	0.013
Hydrogen	0.000005	2.02	0
Argon	0.00933	39.94	0.373
Neon	0.000018	20.18	0
Helium	0.000005	4.00	0
Krypton	0.00001	83.8	0
Xenon	0.09 10-6	131.29	0
Total Molecular Mass of dry Air			28.97

Water Vapor

- Water vapor H2O one Oxygen atom and two Hydrogen atoms
- Hydrogen 1 atomic unit
- Oxygen 16 atomic units
- Water vapor 18 atomic units
- Ratio of wet air to dry air
- Dry air is more dense than humid air!

$$\frac{M_v}{M_d} = \frac{18}{28.97} = 0.622$$

Atmospheric Moisture

Vapor pressure

- e = vapor pressure of water vapor
- Water vapor normally behaves as an ideal gas
- \square ρ_v = vapor density (mass per unit volume)
- T = Temperature (degK)
- R_v = vapor gas constant = R_o/M_v
- $-R_0$ = Universal gas constant
- $-M_{v}$ = molecular weight of water vapor
- <u>Partial pressure</u> of water (vapor pressure) adds to partial pressures of the other gaseous constituents

 $e = \rho_{\nu} R_{\nu} T$

- Water vapor is about 1-2% of total pressure
- Humidity quantity of water vapor present in air (absolute, specific or a relative value)
- Specific humidity ratio of mass of water vapor in moist air to mass of air

Specific Humidity, q_v

- $q_v = \text{mass of water vapor per mass}$ of moist air
- $M_v = \text{mass of water vapor}$
- M_d = mass of dry air

$$q_{v} = \frac{M_{v}}{M_{v} + M_{d}} = \frac{\frac{m_{v}e}{RT}}{\frac{m_{v}e}{RT} + \frac{m_{d}P_{d}}{RT}} = \frac{m_{v}e}{m_{v}e + m_{d}P_{d}}$$

$$= \frac{\frac{m_{v}}{m_{d}}e}{\frac{m_{v}}{m_{d}}e + (P - e)} = \frac{0.622e}{0.622e + (P - e)} = \frac{0.622e}{P - 0.378e}$$

$$\approx \frac{0.622e}{P}$$

- $m_v =$ molecular wt of water vapor
- m_d = molecular wt of dry air
- *e* = partial pressure of water vapor
- P_d = partial pressure of the dry air
- *P* = Total pressure of the air

$$q_{v} = \frac{\rho_{v}}{\rho_{a}} \approx \frac{0.622e}{P}$$

Relative Humidity, R_h

- e_s = Saturation Vapor Pressure
 - Max moisture air can hold @ given temp

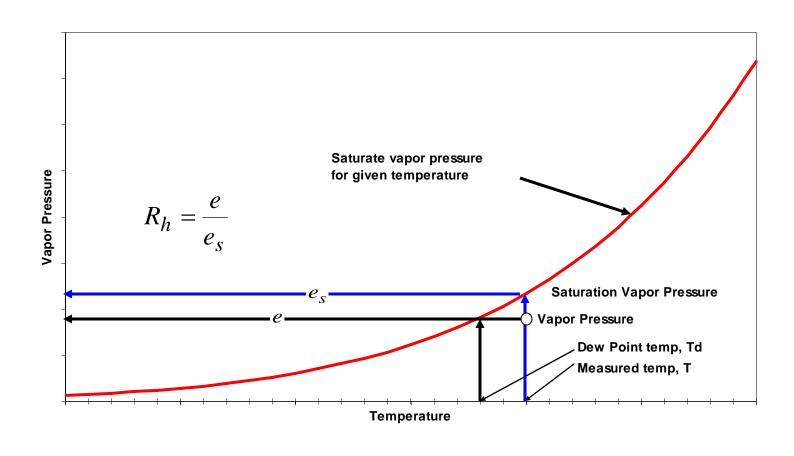
$$e_{S} = 611e^{17.27T/237.3T}$$

• R_h = vapor press/sat. vapor press

$$R_h = \frac{e}{e_s}$$

- T_d = Dew Point Temperature
 - temp at which air becomes saturated

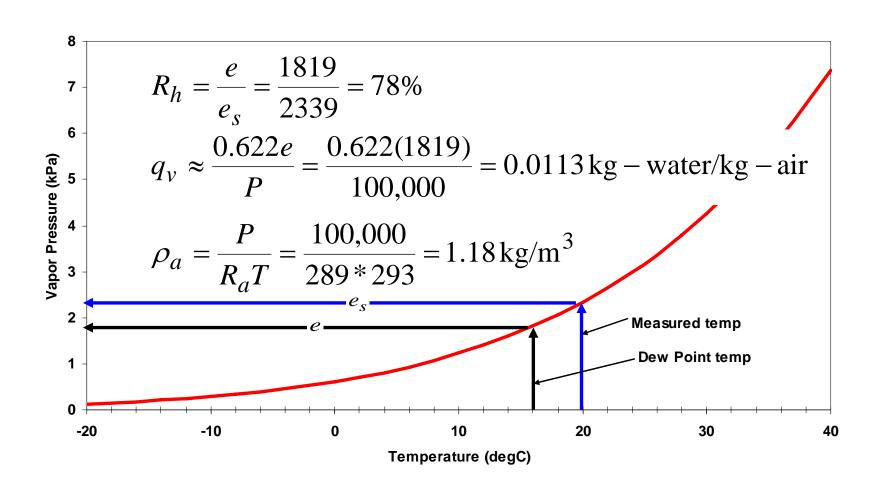
Even More Humidity



Example

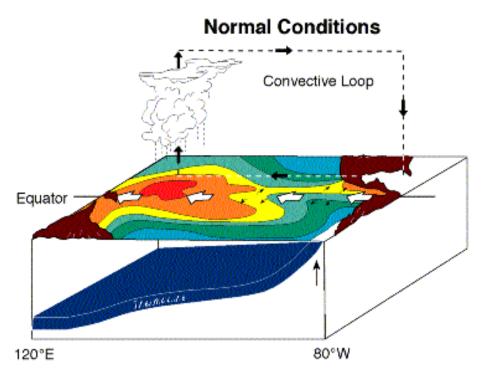
- Air pressure = 100 kPa
- Air temperature = 20 degC
- Wet-bulb (Dew Point) temperature = 16 degC
- Find:
 - Vapor pressure
 - Relative humidity,
 - Specific humidity,
 - Air density

Example



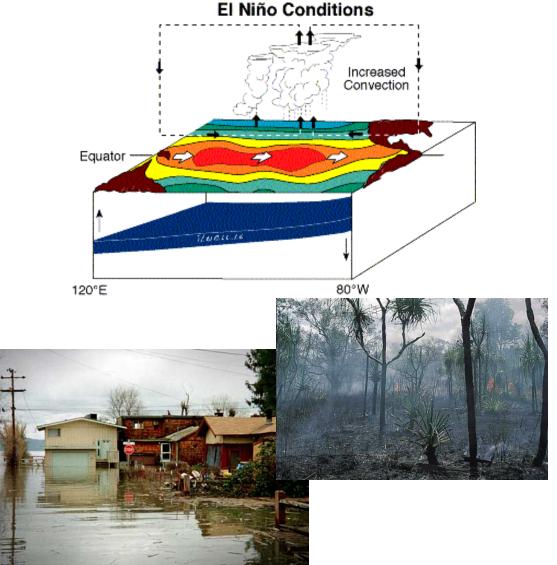
Non - El Nino Conditions

- Trade winds blow west across the Pacific,
- Piling up warm water in the west Pacific.
- Surface temperature is warmer in the west, and cooler off South America, due to upwelling of cold water from deeper levels.
- This cold water is nutrient-rich, supporting high levels of primary productivity, diverse marine ecosystems, and major fisheries.
- Rainfall is over the warmest water, and the east Pacific is relatively dry.



El Nino Conditions

- Trade winds relax in the central and western Pacific
- Rise in sea surface temperature
- Decline in primary productivity, adversely affecting higher levels of the food chain, including commercial fisheries.
- Rainfall follows the warm water eastward, with associated flooding in Peru and drought in Indonesia and Australia.



Risk from El Nino

